Communications

Subglutinols A and B: Immunosuppressive **Compounds from the Endophytic Fungus** Fusarium subglutinans

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Immunosuppressive drugs are used today to prevent allograft rejection in transplant patients, and in the future they could be used to treat autoimmune diseases such as rheumatoid arthritis and insulin-dependent diabetes.¹ Currently approved immunosuppressive agents such as cyclosporin A and FK506 possess some undesirable side effects, and the search for better immunosuppressive agents continues.² We have been investigating endophytic fungi-the fungi that live in the intercellular spaces inside living plants-as a source of novel compounds. This search has emphasized fungi from plants with demonstrated biological activity, and in screening the fungi from the perennial twining vine Tripterygium wilfordii, we discovered two novel immunosuppressive compounds that form the basis of this report. T. wilfordii, which has been used in Chinese traditional medicine for over two thousand years, is familiar to chemists as the source of cytotoxic diterpene lactones such as tripdiolide³ and insecticidal alkaloids⁴ such as wilfordine. Recently, several groups^{5,6} have reported the isolation of immunosuppressive triterpenes from T. wilfordii. An endophytic fungus from T. wilfordii, identified as Fusar*ium subglutinans*, produces the immunosuppressive but noncytotoxic diterpene pyrones subglutinol A (1) and B (2).

F. subglutinans was cultured in modified M-1-D medium, and immunosuppressive activity, as judged by the mixed lymphocyte reaction, appeared in the ethyl acetate extract of the culture medium. The extract was chromatographed on silica gel using CH₂Cl₂-MeOH mixtures. The active fractions were subjected to reverse phase HPLC on a C-18 column with a solvent gradient of 75% MeOH-H₂O to 100% MeOH to give two active compounds, subglutinol A (1) and B (2) (3:1). Compounds 1 (mp 185 °C dec; $[\alpha]^{25}_{D} = -58.7^{\circ}$ (c 0.027, MeOH) and 2

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	Table 1.	NMR Data for Subglutinol A (1) in MeOD		
	¹³ C ^a	¹ H ^{b,c}	HMBC d	ROESY e
1	26.2 t	1.56 (qd, 13.0, 5.0), 1.45	2	
2	32.1 t	2.45 (br td, 13.5, 6.0),	1, 4, 19	
		2.05 (br dd, 14.0, 3.5)		
3	149.9 s		2, 4, 20	
4	56.5 d	2.17 (dd, 11.5, 4.5)	2, 18, 19	0.96, 4.25
5	39.6 s		4, 18	
6	35.9 t	1.90	7, 18	
		1.38 (dt, 14.0, 3.5)		
7	23.3 t	1.78, 1.74		0.85, 0.96
8	88.6 d	3.17 (dd, 11.5, 4.0)	11, 17	1.69, 4.75
9	$45.5 \mathrm{s}$		11, 17	
10	46.5 d	1.74	2, 4, 6, 17, 18	
11	49.0 t	1.90, 1.45 (dd, 11.5, 4.0)	8, 17	
12	75.9 d	4.75 (td, 9.5, 3.5)	11	1.69, 3.17
13	128.4 d	5.35 (br dt, 8.5, 1.5)	11, 15, 16	1.72, 0.85
14	136.4 s		15, 16	
15	25.8 q	1.72 (d, 0.8)	13, 16	5.35
16	18.2 q	1.69 (d, 1.0)	13, 15	3.17
17	17.5 q	0.85 (s)	8, 11	0.96, 5.35
18	25.1 q	0.96 (s)	4, 6, 10	0.85, 2.17,
				4.25
19	110.7 t	4.50 (t, 2.5), 4.25 (t, 2.0)	2, 4	0.96, 2.17
20	22.7 t	2.74 (dd, 13.5, 12.0),		
		2.58 (dd, 13.5, 5.0)		
21	103.7 s		20	
22*	165.35 s		20, 27	
23	109.0 s		26, 27	
24	156.8 s		26, 27	
25*	165.30 s		20	
26	17.3 q	2.19 (s)		
27	10.4 q	1.91 (s)		

 a $^{13}\mathrm{C}$ spectrum was recorded at 100 MHz, referenced to d_3 -MeOD at 49.0 ppm. Multiplicity was based on the DEPT spectrum. ^b ¹H spectrum was recorded at 500 MHz, referenced to CHD₂OD at 3.30 ppm. Assignments to carbons were based on HMQC data. ^c Those signals without indicated splitting patterns are either multiplets or buried within other signals. ^d Numbers correspond to position of the proton. e The chemical shifts of the protons are given. Only those that are relevant to stereochemical assignment are shown.



(mp 220 °C dec; $[\alpha]^{25}_{D} = -75.0^{\circ}$ (c 0.288, MeOH) have identical UV spectra, very similar ¹H- and ¹³C-NMR spectra, and the same molecular formula, C₂₇H₃₈O₄, as judged by HRFABMS. The formula requires nine degrees of unsaturation, and since nine signals in the ¹³C-NMR are sp², the compounds contain four rings and a carbonyl (Table 1). A broad IR band at 3400-3100 cm⁻¹ and a strong absorption at 1664 cm⁻¹ indicate an alcohol, which is consistent with the exchangeable proton in the ¹H-NMR, and a conjugated carbonyl.

The planar structure of the more abundant subglutinol A (1) can be deduced from analysis of one- and twodimensional NMR spectra (Table 1). Partial structures

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Figure 1. Fragments of 1 derived from COSY and HMBC spectra.

A, B, and C for 1 (Figure 1) are obtained from analyses of the COSY, HMQC, and HMBC spectra as sketched below. In A, the position of the exocyclic double bond is determined from the heteronuclear correlations observed between the protons at δ 2.05 (C-2) and 2.17 (C-4) to the carbons at δ 149.9 (C-3) and 110.7 (C-23). Because the protons of C-1 and C-2 are coupled, the methine signal at δ 46.5 (C-10) must have three-bond correlations to the protons at δ 2.05 (C-2), 2.17 (C-4), 0.85 (C-17), and 0.96 (C-18). The closure of ring A is established from the observed cross peaks of the signal at δ 39.6 (C-5) to the $\delta 2.17$ (C-4) and 0.96 (C-18) resonances. In **B**, the olefinic signal at δ 5.35 (C-13) shows a large coupling to δ 4.75 (C-12) and long-range coupling to the terminal methyl groups. Since the quaternary carbon signal at δ 45.5 (C-9) correlates to proton signals at δ 3.17 (C-8), 1.90 (C-11), and 1.45 (C-1), and the carbon signal at δ 35.9 (C-6) correlates to the proton signal at δ 1.78 (C-7), A and B are connected to form the diterpene skeleton of 1.

The remaining carbon resonances consist of five sp², one methylene, and two methyl signals. Fragment C is partially constructed by signals at δ 109.0 and 156.8 having cross peaks to both methyl signals. Further analysis of the HMBC spectra indicate that the three remaining sp² signals must be arranged to form a tautomeric α - or γ -pyrone, since the methylene protons at δ 2.74 and 2.58 correlate with all three carbon resonances. Since λ_{max} for 1 (293 nm) is significantly closer to the expected α -pyrone value,⁷ and there is no literature precedent for a free 2-hydroxy-4-(γ)-pyrone, 1 is drawn as the α -pyrone tautomer.

Finally, fragment **B** must contain a pentacyclic ether ring. Although no heteronuclear correlation is observed between the C-12 and C-8 positions, a through-space correlation is observed in the ROESY spectrum of **1** (Figure 2). This NOE enhancement determines the relative stereochemistry at C-8 and C-12 and forms the basis of the relative configurations at all of the other chiral centers as illustrated in Figure 2. NOE experiments indicate that subglutinol B (**2**) is the C-12 epimer of subglutinol A (**1**) (Figure 2). A final check on the assigned stereochemistry is provided by a single crystal X-ray diffraction analysis of **2**, which finally crystallized after the NMR experiments on **1** and **2** were completed.⁸ The results of this analysis are shown in Figure 3. In



Figure 2. NOE correlations of 1 and 2.



Figure 3. A computer-generated perspective drawing of the final X-ray model of subglutinol B (2).

the solid state, compound **2** clearly exists as the α -pyrone, not the γ -pyrone, tautomer.

The current analysis defines only the relative stereochemistry of compounds 1 and 2. The absolute configurations shown are drawn to agree with those established for the related compounds viridoxin A and B, antiinsecticidal metabolites of the entomopathogenic fungus *Metarhizium flavoviride*.⁹

Compounds 1 and 2 are equipotent in the mixed lymphocyte reaction (MLR) assay and thymocyte proliferation (TP) assay (IC₅₀ 0.1 μ M), and this equipotency suggests that the residue at C-12 does not interact with the biological target. In the same assay systems, cyclosporin A is roughly as potent in the MLR assay and 10⁴ more potent in the TP assay, but the lack of toxicity with 1 and 2 suggests that they definitely should be explored in greater detail. Further work on the *in vivo* activity of subglutinol A (1) and B (2) is in progress and will be reported in due course.

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Supporting Information Available: Spectral and physical data for subglutinol A (1) and B (2) (8 pages).

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⁽⁸⁾ Space group $P2_12_12_1$ with a = 7.187(1), b = 9.685(2), and c = 34.405(7) Å and one molecule in the asymmetric unit. The final *R*-factor is 0.075 for the 2σ data set.

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